Chapter 3: conductivity and conductometry

ACTIVITY 4 : Salinity of sea water

Objective: using a calibration curve in order to determine an unknown concentration

Introduction

The first step in using conductivity to determine salinity is to quantify the relationship between conductivity and salinity. In other words, we want to know exactly how conductivity changes when salinity changes.

The best way to do this is to use samples with known salinities and measure their conductivity with a conductimeter (or salinometer). Making conductivity measurements on samples with a known salinity, oceanographers can determine the salinity on unknown samples. Oceanographers typically use commercially available standardized seawater (with a precisely measured salinity) to calibrate their salinometers.

Table 1 below illustrates typical conductivity of various samples with a known salinity at 20 °C and atmospheric pressure.

Source: http://hrsbstaff.ednet.ns.ca/vturner/handouts%20Oceans%2011/Salinity%20and%20Conductivity.pdf

Table 1: Conductivity measurements of standard solutions

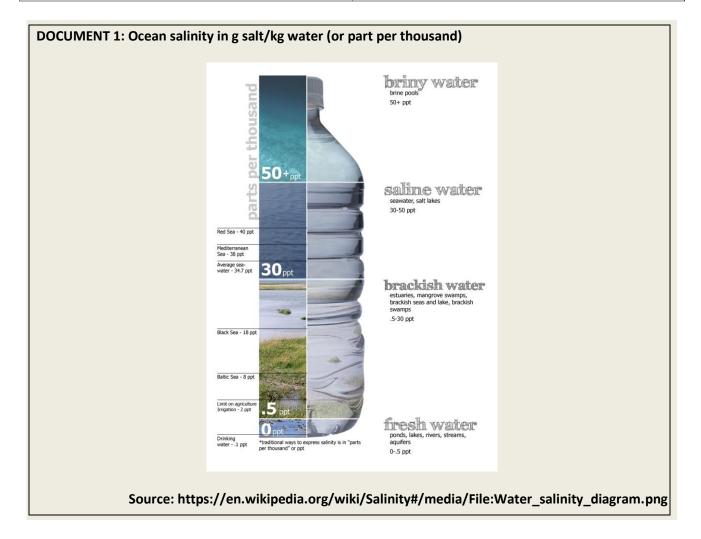
Source: http://hrsbstaff.ednet.ns.ca/vturner/handouts%20Oceans%2011/Salinity%20and%20Conductivity.pdf

Salinity (g salt/kg water)	Conductivity (mS/cm)
25.6	40.0
27.4	42.5
28.4	44
29.9	46
31.7	48.5
33.2	50.5
34.3	52
35.8	54
38	57
39.5	59

Table 2: Conductivity measurements of unknown solutions

Source: http://hrsbstaff.ednet.ns.ca/vturner/handouts%20Oceans%2011/Salinity%20and%20Conductivity.pdf

Sample	Conductivity (mS/cm)
1	41
2	45.5
3	47
4	50



Using a calibration curve

Using all documents, determine the salinity of the four samples in Table 2. Determine which category of water they belong to: fresh water (drinkable), brackish water, saline water, briny water?

Activity summary

What you must remember:

- conductivity
- standard solution
- calibration curve

Skills linked to the curriculum:

Compétences	Capacités à maitriser
– ANA	Interpréter ou prévoir l'allure d'une courbe de titrage conductimétrique à partir de données, sans tenir compte de l'effet de la dilution.
– REA	Réaliser un protocole de titrage mettant en jeu une réaction suivie par conductimétrie.
– COM	Formuler et argumenter des réponses structurées Formuler et présenter une conclusion